Aesthetic Rehabilitation of Discoloured Nonvital Anterior tooth with Carbamide Peroxide Bleaching: Case Series

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ABSTRACT
Discolouration of teeth, especially the anteriores, can result in considerably cosmetic impairment in person. Combine effects of intrinsic and extrinsic colour determines the appearance of teeth. Whitening of teeth with bleaching is a more conservative therapeutic method than full crowns, veneers or composite restorations which is more invasive and expensive. Among bleaching techniques, in office bleaching with carbamide peroxide provide superior aesthetic result in short period of time with no adverse effects. This paper presents case series of tooth discolouration in non-vital tooth which was successfully bleached using 35 % carbamide peroxide. After 1 year follow up the prognosis was good with no reversal of tooth discolouration. This case report allows the better understanding of the concept of nonvital tooth bleaching with carbamide peroxide which gives a non-invasive alternative for aesthetic purpose in preserving the natural tooth structure.

CASE 1
A 35 year old female patient was reported to the department of conservative dentistry and endodontics with complaint of discoloured upper anterior teeth. Patient had history of trauma with anterior teeth 5 years back. Intraoral examination revealed brown discoloured maxillary right central incisor [Table/Fig–1]. An intraoral radiograph showed large periapical lesion with 11 [Table/Fig–2]. Periapical cyst was diagnosed with 11. Nonsurgical management with endodontic treatment and in office bleaching was planned. Access opening and complete biomechanical preparation was completed. Calcium hydroxide (RC Cal; Prime Dental Products, Thane, India.) intracanal medicament was placed and recalled the patient after three weeks. Obturation with lateral condensation was done and recalled after seven days for bleaching procedure. Gutta-percha filling was removed from the access cavity to 2 mm bellow cemento-enamel juction with gates glidden drill (Dentsply Maillefer, Ballaigues, Switzerland). A plug of resin modified glass ionomer (GC Fuji PLUS, GC Corporation, Tokyo, Japan.) was placed over the gutta-percha in cervical region to prevent percolation of bleaching agent in apical area. Opalescence gel (Ultradent, St. Louis, MO, USA), containing 35% carbamide peroxide was used for bleaching.

OpalDam was applied and cure for 20 second over gingiva as well as 2 mm over the adjacent the tooth for protection of gingiva. Opalescence gel was applied inside the pulp chamber as well as over labial surface and kept for 30 minutes [Table/Fig–3]. Material was removed from the tooth and thoroughly rinsed with water. The access cavity was sealed with IRM cement (Dentsply De Trey

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GmbH, Konstanz, Germany) and patient was recalled after 72 hours for evaluation. Same procedure was performed in second visit as slight discolouration was present with 11. Two sittings of bleaching procedure showed complete removal of stains from tooth and present colour was comparable to adjacent tooth [Table/Fig-4]. Coronal access cavity was sealed with composite resin (3M Dental Products, St Paul, MN, USA). One year evaluation of patient showed complete healing of periapical lesion and no significant shade regression was observed [Table/Fig-5a and 5b].

**CASE 2**

A 29-year-old female patient was reported to the department of conservative dentistry and endodontics with complaint of discoloured upper anterior teeth. Patient had history of trauma with anterior teeth 3 years back and she had taken root canal treatment for the same. Intraoral examination revealed brown discoloured maxillary left central incisor [Table/Fig-6]. An intraoral radiograph showed obturation and normal periapical tissue with 21. In office bleaching with 35 % carbamide peroxide (Opalescence gel, Ultradent, St. Louis, MO, USA) was decided for this patient and similar bleaching procedure was performed as described in above case. Tooth showed significant changes in shade after 72 hours [Table/Fig-7].
incisor one year back. Patient had taken root canal treatment with the same tooth [Table/Fig–8a]. Single visit of 35% carbamide peroxide (Opalescence gel, Ultradent, St. Louis, MO, USA) in office bleaching showed significant changes in shade of 11 [Table/Fig–8b]. There was no significant shade reversal observed in both the cases i.e. 2nd & 3rd, after three months of follow up period.

DISCUSSION
Bleaching of discoloured nonvital teeth was first described during the middle of the 19th century advocating different chemical agents [1]. An early description of the use of hydrogen peroxide for bleaching was reported by Harlan [2]. Action of hydrogen peroxide is mainly due to reactive oxygen through the formation of free radicals [3].

Thermocatalytic technique was used to bleach nonvital teeth as the application of heat increases the bleaching properties of hydrogen peroxide by accelerating the bleaching reaction [1]. Today the thermocatalytic technique and high concentration of hydrogen peroxide is used less due to increase risk of external cervical resorption [4]. Exact mechanism of resorption in bleached tooth has not yet been adequately explained. There is speculation that hydrogen peroxide diffuses through dentinal tubules, cementum, periodontal ligament and reach to periradicular bone and directly induces an inflammatory resorption process [5]. There are different predisposing factor for diffusion of hydrogen peroxide to cervical tissue. Patients those had bleaching therapy at young age [6], application of heat (thermocatalytic method) [7] and natural anatomic defects at CEJ also responsible for diffusion of hydrogen peroxide into periradicular tissue [8]. Also the sealing of root filling with 2mm layer of glass ionomer cement at CEJ is essential to prevent radicular penetration of bleaching agents [9].

An alternative to thermocatalytic bleaching is the walking bleach technique with sodium perborate–hydrogen peroxide solution which did not show any cervical resorption even after 1 year [4]. More recently carbamide peroxide is the choice of material for intracoronal bleaching due to low levels of extraradicular diffusion and its effectiveness [10].

In present cases, 35% carbamide peroxide was used as a bleaching material and placed on labial surface as well as in the pulp chamber because it is recommended that pulp chamber should accessible during bleaching procedure for proper penetration of material into discoloured tooth [1]. High concentration of bleaching agent promote faster tooth bleaching than lower concentrations of carbamide peroxide which takes longer time to bleach teeth but eventually achieve the same results as higher concentrations [11, 12]. Carrasco, et al., [13] reported that 37% carbamide peroxide as an intracoronal bleaching increased dentinal permeability, which was not observed for 27% carbamide peroxide.

35% Carbamide peroxide when come in contact with water breaks into 23–25% of urea and 10–12% of hydrogen peroxide [14]. Urea is capable of penetrating enamel and affecting the interprismatic region [15]. It is also capable of attacking the protein structure and denatures it. Structural alterations may occur in enamel proteins such as enamelin and amelogenin, thus urea increases the permeability of enamel to peroxides and free radicals [16]. Initial faster rate of bleaching was observed with higher concentration of carbamide peroxide [17]. The efficacy of hydrogen peroxide containing products are approximately the same when compared with carbamide peroxide containing products with equivalent or similar hydrogen peroxide content either tested in–vivo [18] or in –vitro [19], Lim MY et al., in their study found that 35% carbamide peroxide and 35% hydrogen peroxide were equally effective for intracoronal bleaching [20].

The effect of carbamide peroxide on enamel surface after bleaching was studied by investigators and found there is no alteration of the surface texture of enamel or slight increase in enamel porosity immediately following treatment which is reversed after 6 months as observed in follow up examination [21,22]. Lewinstein et al., observed a slight reduction in enamel surface microhardness following 35% carbamide peroxide treatment on human enamel which was completely reversed when treated with 0.05% fluoride solution [23]. Carbamide peroxide was effective against mild to moderate fluorotic stains but more time is required to assimilate with the normal enamel in surrounding, due to different responses of sound and fluorotic enamel to the bleaching regimes [24]. Single bleaching therapy seems to be satisfying nonabrasive approach in cases of mild or moderate fluorosis [24].

Thus, non–vital tooth office bleaching with carbamide peroxide is a relatively fast method in whitening the discoloured tooth and equally effective to hydrogen peroxide with less complications of external cervical resorption.

CONCLUSION
From the present cases it was concluded that, 35% carbamide peroxide was effective as in office bleaching of discoloured nonvital teeth with significant improvement in colour of teeth in relatively short working time. Also the bleaching procedure is more economical and less invasive than other invasive tooth whitening procedures. This technique provides minimum risk of complications and more promising results.

REFERENCES
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